



## A practical introduction to the simulation of molecular systems

By-

2015. Soft cover. Book Condition: New. 1st Edition. Preface to the first editionPreface to the second edition1 Preliminaries1.1 Introduction 1.2 Python 1.3 Object-oriented programming 1.4 The pDynamo library 1.5 Notatio and units 2 Chemical models and representations2.1 Introduction2.2 The System class2.3 Example 12.4 Commomolecular representations 2.5 Example 23 Coordinates and coordinate manipulations 3.1 Introduction 3.2 Connectivity 3.3 Internal coordinates 3.4 Example 33.5 Miscellaneous transformations 3.6 Superimposing structures 3.7 Example 44 Quantum chemical models4.1 Introduction4.2 The Born-Oppenheimer approximation 4.3 Strategies for obtaining energies oa potential energy surface4.4 Molecular orbital methods4.5 The Hartree-Fock approximation4.6 Analysis of the charge density 4.7 Example 54.8 Derivatives of the potential energy4.9 Example 65 Molecular mechanics5.1 Introduction5.2 Typical empirical energy functions 5.3 Calculating a molecular mechanics energy5.4 Example 75.5 Parametrizing potential energy functions 5.6 Soft constraints Hybrid potentials 6.1 Introduction6.2 Combining QC and MM potentials6.3 Example 86.4 Covalent bonds betweeQC and MM atoms6.5 Example 9Finding stationary points and reactiopaths opotentialenergy surfaces7.1 Introduction7.2 Exploring potential energy surfaces 7.3 Locating minima 7.4 Example 107.5 Locating saddle points7.6 Example 117.7 Following reactiopaths7.8 Example 127.9 Determining complete reactiopaths7.10 Example 138 Normal mode analysis8.1 Introduction8.2 Calculatioof the normal modes8.3 Rotational and translational modes8.4

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